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### 内容提要

本书精选了 331 个有机人名反应。每个反应均给出一步一步详尽的电子转移机理过程和相关原始的及最新含 2003 年度在内的参考文献。全书列出的 2740 多篇参考文献中有相当部分是综述 类论文,以帮助读者更好地理解和认识有机反应,同时为深入应用有机反应提供了方便。

本中译本是根据 2003 年出版的第 2 版翻译的,可供大学院校师生和有关科研工作者参考使用。

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## 第二版前言

第二版有五个方面的改进:1)增加了16个人名反应。2)改正了初版中的打印错误和几个有疑点的机理。我想感谢 Massachuetts Institute of Technology 的 Danheiser R. L. 教授和 Michigan State University 的 Lian Y. 先生,他们提出了极富价值的评论和建议。我还要向 California Institute of Technology 的 Stoltz B. M. 教授及他的学生 Ashley E. . Behenna D. . Caspi D. . Garg N. . Greene B. . May J. . Spessad S. . Tambar U. . Trend R 和 Zeidan R. 致意,他们对第二版的最终稿样进行了校对。3)参考文献得到增加并更为及时。4)提供了更详尽的索引使读者能更方便地检索书本。5)作为一种记忆法,对每个所给出的人名反应作一小段论述看来无论对初学者还是有经验的专业人员来讲都是有益的,因此,大部分反应都有了评述。最后,我很感谢得到允许在内封上使用邮票,而这些邮票的版权仍然是归各个邮政当局所拥有的。

Jack Li

Ann Arbor, Michigan, 2003年5月

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# 译校者的话

有机化学是一门富有个人特色和高度竞争性的学科,化学家们已经发现了难以计数的各类化学反应,其中有相当数量的有机反应是以一个或几个科学家的姓名来归类和予以命名的,有机人名反应可以说是有机化学的一大特色,它们占有有机反应的核心地位。许多有机人名反应的发现者都得到过诺贝尔化学奖。

有机反应被冠以人名,首先是为了纪念首次发现这个(类)反应或是对这个(类)反应作出深人研究取得突出成就的科学家。二是为了方便学习、记忆和运用这个(类)反应。现代有机化学在 19 世纪开始发展起来后,人名反应也就随之产生并一直在不断增加。有机反应能否冠以人名并无严格标准,但通常是与反应的新颖性、重要性、应用性,推广性密切相关的。迄今为止,尽管难以确切统计,但大概已有上千个人名反应曾被提及或应用过,为人们广为熟知的也有几百个。有些人名反应随着时间的推移因种种原因失去其应用价值后逐渐为人们所淡忘,有些人名反应则由于后人的不断改进而更趋完美,所冠的人名也有变化或增加。毫无疑问的是,要学好有机化学,熟悉人名反应是一个基本要求,要做好有机化学,掌握更多的人名反应是一个素质要求。

由 Li J. J. 编著的《Name Reactions A Collection of Detailed Reaction Mechanisms》是由德国的 Springer 公司 2002 年出版发行的,2003 年 8 月又出了第 2 版。本中译本就是根据第 2 版翻译而成的。国内外涉及有机人名反应的著作也有一些,但本书是颇有特色的一种。它并不追求齐全,但富有时代感,着眼于反应是否创新及有应用价值。全书收集了 331 个有机人名反应(少数几个是人名试剂,个别的是俗称而非人名)。其中发现于 19世纪的有 60 多个,最早的一个是 1838 年由现代有机化学的开拓者 Liebig J. 所发表的二苯乙醇酸重排反应(Benzilic acid rearrangement)。 20世纪前半叶的有近 120 个,20 世纪 80 年代以后发表的有 40 多个,最新的一个是2000 年发表的烷基硼酸和硫醇酯在 Pd 催化下偶联成酮的反应(Liebeskind-Srogl coupling)。每个反应均给出详尽而又完整的一步一步电子转移的机理过程。同时,全书还列出了 2740 多篇参考文献。每个反应基本上都附有原始发表的论文,其余的则是最近的进展和研究性、综述性论文,不少文献是近年包括 2003 年才发表的。这样的编写方式对读者深入了解和

应用某个(类)反应是很有帮助并极富参考价值的。译校工作对英文原著中的一些编校差错作了改正,英文人名未作翻译。我们希望本书中文版能成为我国有机化学工作者和学习者身边常备常用和不可或缺的一本工具参考书。

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于上海. 2003年8月

# www.terijet, Athania

## 前 言

什么是名字?我们叫做玫瑰的一种花,要是换了个名字,它的香味还是同样地芬芳"。然而,与莎士比亚所下的定义不同,有机化学中的人名反应及其机理就其深远的应用内容及对有机反应的影响来说都是极其迷人的。而理解这些机理更可大大增强我们解决复杂化学问题的能力。事实上,某些人名反应,如 Barton-McCombie 反应<sup>21</sup>,也就是较好地理解了反应机理后的直接结果。另外,认识反应是怎样进行的有助于判断反应不能给出所需产物或导致副反应、副产物的过程,机理可以提供反应在哪里出了偏差的线索。

当我还是一个研究生时我就已经开始搜集人名的和非人名的有机反应及其机理信息。我也想到,我的许多同事也在做着同样的事情,这些努力可以更方便地通过一个列出有意义的及有用的人名反应机理的专题来做到。到目前为止,我已经收集到最近的当代人名反应并增加了更及时的参考文献,特别是那些最新的综述论文。为反映出不对称合成的进展,该领域的人名反应已经全部收集起来。书中描绘的大部分机理细节是不喻自明的,因此不再给予详细的文字叙述,但在某些重要的转化类型中必需的专用术语还是着重提及。某些反应给出了一些简短的易记的注释,但它们并非是精确的定义。参考文献中的第一篇一般是原始论文,其余的是文章或综述论文。对人名反应有更多兴趣的读者可以参阅这些参考文献和有关书籍3~70。

我要感谢 Wayne State University 的 Brian J. Myers 博士, Indiana University 的 Jeffery J. Johnson 教授和 Bernard College 的 Christian M. Rojas 教授,他们阅读了样稿并且提供了许多有价值的评论和意见。特别感谢 Dartmouth College 的 Gordon W. Gribble 教授、Colorado State University 的 Louis S. Hegedus 教授和 University of Minnesota 的 Thomas R. Hoye 教授,他们对草稿提出了评议。此外,我非常感激 Nadia M. Ahmad、John (Jack) Hodges、Michael D Kaufman、Peter L. Toogood 和 Kim E. Werner 等人,他们进行了校对样稿的工作。当然,本书的任何差错,都是由我而造成的。我还想要感谢 Merck & Co., Inc. 的 Ann Smith 小姐,她给予了有益的讨论和交流。最后但同样重要的是,我要感激我的妻子 Sherry Cun—hua Cai,她对本书的整个编写工作表达了理解和支持。

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# 缩写词和首字母缩写词

Ac (乙酰基)

AIBN 2.2'-azobisisobutyronitrile (偶氮双异丁腈)

Alpine-borane $^{\circledast}$  B-isopinocamphenyl-9-borabicyclo [3, 3, 1]-nonane (B-

3α-蒎烯-9-BBN) (水溶液)

B (碱)

9-BBN 9-borabicyclo[3, 3, 1]nonane (9-硼-双环[3, 3, 1])壬烷

BINAP 2.2-bis(diphenylphosphino)-1,1'-binaphthyl (2,2'-双二

苯膦-1'1-联萘)

Boc tert-butyloxycarbonyl (叔丁氧羰基)

t-Bu tert-buty! (叔丁基)

Cat 催化(剂)

Cbz benzyloxycarbonyl (苄氧羰基)

conc (液)

m-CPBA m-chloroperoxybenzoic acid (间氯过氧苯甲酸)

CuTC copper thiophene-2-carboxylate (噻吩-2-甲酸铜)

DABCO 1,4-diazabicyclo[2.2.2]octane (1,4-二氮双环[2.2.2]辛

烷)

dba dibenzylideneacetone (二亚苄基丙酮)

DBU 1.8-diazabicyclo [5.4.0.] undec-7-ene(1.8-二氮双环

[5.4.0]十一碳-7)-烯

DCC 1.3-dicyclohexylcarbodiimide (1,3-双环己基碳二亚胺)

DDQ 2,3-dichloro-5,6-dicyano-1,4-benzoquinone (2,3-1,5-

5,6-二氟-1,4-苯醌)

DEAD diethyl azodicarboxylate (偶氮二甲酸二乙酯)

Δ (回流下加熱)

(DHQ)2-PHAL 1,4-bis(9-O-dihydroquinine)-phthalazine (1,4-双(9-O-

二氢奎宁>-2,3-二氮杂萘)

(DHQD)<sub>2</sub>-PHAL 1,4-bis(9-O-dihydroquinidine)-phthalazine (1,4-双(9-O-

二氢奎尼啶>-2,3-二氮杂萘)

DIBAL diisobutylaluminum hydride (二异丁基氢化铝)

DMA N, N-dimethylacetamide (N, N-二甲基乙酰胺)

www.cheenich. Arts felt fil N,N-dimethylaminopyridine (N,N-二甲胺基吡啶) 1,2-dimethoxyethane (1,2-二甲氧基乙烷) dimethyformamide (二甲基甲酰胺) DMF dimethylsulfide (二甲基硫醚) DMS dimethylsulfoxide (二甲亚砜) DMSO dimethylsulfoxonium methylide (二甲基氧化锍亚甲基) DMSY dimethoxytrityl (二甲氧基三苯甲基自由基) **DMT** 1,4-bis(diphenylphosphino)butane (1,4-双〈二苯基膦〉 dppb 干烷) dppe 1,2-bis(diphenylphosphino)ethane (1,2-双〈二苯基膦〉 乙烷) 1.1'-bis(diphenylphosphino)ferrocene (1.1'-双(二苯基 dppf 膦>二茂铁> dppp 1,3-bis(diphenylphosphino)propane (1,3-双(二苯基膦) 丙烷) E1(单分子消除) Elcb (碱诱导下经碳负离子而进行的二步的 &消除反应) E2(双分子消除) (对映过量) ee endo (内向) Eq (等当量的) (外向) exo **HMPA** hexamethyphosphoric triamide (六甲基膦酰三胺) Imd imidazole(咪唑) LAH lithium aluminum-hydride (锂铝氢) LDA lithium diisopropylamide (二异丙基锂铵) lithium hexamethyldisilazane (六甲基二硅胺锂) LHMDS LTMP lithium 2,2,6,6-tetramethylpiperidine (2,2,6,6-四甲基 哌啶锂)

MVK methyl vinyl ketone (甲基乙烯基酮) N-bromosuccinimide (N-溴代丁二酰亚胺) N-chlorosuccinimide (N-氯代丁二酰亚胺)

N-iodosuccinimide (N-碘代丁二酰亚胺) NIS

mestyl (甲磺酰基)

Mes

**NBS** 

NCS

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1-methyl-2-pyrrolidinone (1-甲基-2-吡咯酮)

Nu (亲核物种)

PCC pyridinium chlorochromate (氯铬酸吡啶鎓盐)

PDC pyridinium dichromate (吡啶重铬酸吡啶鎓盐)

Py (pyridine) (吡啶)

ref (回流)

SET single electron transfer (单电子转移)

SvAr nucleophille substitution on an aromatic ring (芳香环上

的亲核取代)

S<sub>N</sub>1 (单分子亲核取代)

S<sub>N</sub>2 (双分子亲核取代)

solv (溶剂)

TBAF tetrabutylammonium fluoride (四丁基氯化铵)

TBDMStert-butyldimethylsilyl(极丁基二甲基硅基)TBStert-butyldimethylsilyl(极丁基二甲基硅基)

Tf trifluoromethanesulfonyl(triflyl)(三氟甲磺酰基)

TFA trifluoroacetic acid (三氟乙酸)

TFAA trifluoroacetic anhydride (三氟乙酸酐)
TFP tri-o-furylphosphine (三(2-呋喃基)膦)

THF tetrahydrofuran (四氢呋喃)

TIPS triisopropylsilyl (三异丙基硅基)

N'-四甲基乙二胺)

TMP tetramethylpiperidine (四甲基哌啶)

TMS trimethylsilyl (三甲基硅基)

Tol toluene or tolyl (甲苯或甲苯基(CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>-))

Tol-BINAP 2,2'-bis(di-p-tolylphosphino)-1,1'-binaphthyl (2,2'-双

〈二对甲苯基膦〉-1,1′-联萘)

Ts(tosyl) tosylate(对甲苯磺酰基)

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正常 Claisen 重排产物进一步重排使  $\beta$ -碳原子与环相连。

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# Aldol 缩合

羰基和一个烯醇负离子或一个烯醇的缩合。一个简单的例子为一个烯醇负离子对一个醛(aldehyde)加成给出一个醇(alcohol),故命名为aldol。

$$R$$
  $R^2$   $R^3$   $R^3$   $R^3$   $R^4$   $R^1$   $R^4$   $R^4$ 

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# Arndt-Eistert 同系化(反应)

用重氮甲烷生成增加一个碳的同系物羧酸

$$SOCI_2$$
 O  $1. CH_2N_2$  O  $0. NV$  R O  $0$ 

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### Auwers 反应

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通式:

最富电子的烷基(更多取代的碳)优先迁移。一般迁移规则: 权烷基 > 仲烷基 > 环己基 > 苄基 > 苯基 > 伯烷基 > 甲基 ≫ H

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Bamford-Stevens 反应和 Shapiro 反应有相似的机理。前者用 Na. NaOMe. LiH. NaH. NaNH。等为碱,而后者用烷基锂和格氏试剂等为碱。结果是 Bamford-Stevens 反应实现热力学控制多取代烯烃的合成,而 Shapiro 反应一般给出少取代烯烃的动力学产物。

$$\begin{array}{c}
 & \text{MeO} \\
 & \text{R}^1 \\
 & \text{R}^2 \\
 & \text{H} \\
 & \text{R}^3
\end{array}$$

$$\begin{array}{c}
 & \text{R}^1 \\
 & \text{R}^2 \\
 & \text{H} \\
 & \text{R}^3
\end{array}$$

$$\begin{array}{c}
 & \text{R}^1 \\
 & \text{R}^3
\end{array}$$

$$\begin{array}{c}
 & \text{R}^1 \\
 & \text{R}^3
\end{array}$$

$$\begin{array}{c}
 & \text{R}^1 \\
 & \text{R}^3
\end{array}$$

在质子性溶剂中:

在非质子溶剂中:

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## Bartoli 吲哚合成

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从相关的硫羰基中间体经由自由基开裂发生醇的去氧作用。

$$R^1$$
 S  $R^2$   $R^3$   $R^3$   $R^4$   $R$ 

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## Baylis-Hillman 反应

亦称为 Morita-Baylis-Hillman 反应,有时又被称为 Rauhut-Currier 反应。这是一个在缺电子烯烃和一个亲核碳之间开成碳-碳键的反应。缺电子烯烃包括丙烯酸酯、丙烯腈、乙烯基酮、乙烯基砜和丙烯醛。亲核性碳则可以是醛、α-烷氧羰基酮,醛亚胺和 Michael 反应的受体。

通式:

X = O,  $NR_2$ ,  $EWG = CO_2R$ , COR, CHO, CN,  $SO_2R$ ,  $SO_3R$ ,  $PO(OEt)_2$ ,  $CONR_2$   $CH_2 = CHCO_2Me$ 

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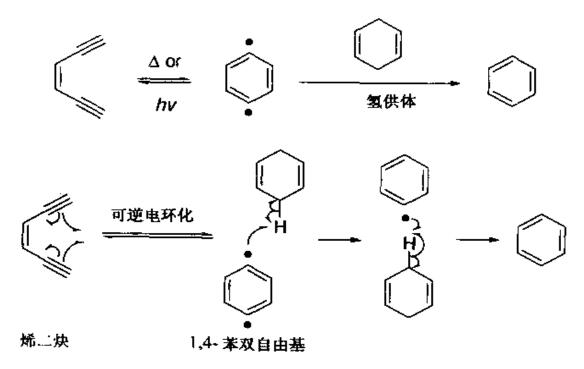
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带供电子基团的苯环:

带吸电子基团的苯环;

$$CO_2H$$
 Na, liq. NH<sub>3</sub>  $CO_2H$   $CO_2$ 

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1,2,4-三唑和亲双烯体(如烯胺)发生杂原子的 **Diels-Alder** 反应再放出 **N**。 而生成吡啶。

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β-卤代烷氧基与锌作用生成烯烃。

$$R$$
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五元杂环热解重排为另一个五元杂环

例<sup>[9]</sup>:

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## Rouveault 醛合成

烷基卤或芳基卤经转化为相应的有机金属化合物后与DMF加成发生甲酰化而得到醛的同系物。

1. M  
2. DMF  

$$R-X \xrightarrow{} R-CHO$$
  
3.  $H^+$   
 $R-X \xrightarrow{} R-M \xrightarrow{} DMF \xrightarrow{} Me_2N \xrightarrow{} O-M \xrightarrow{} H^+ \xrightarrow{} R-CHO$ 

Comins 对此反应有所改进<sup>[6]</sup>:

$$R_{2}N-CHO \xrightarrow{1. R'MgX} R'-CHO$$

$$2. H^{*}$$

$$R_{2}N \xrightarrow{O} R' -CHO + \overline{N}R_{2}$$

$$R'-MgX$$

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烯烃与硼烷加成后所生成的有机硼烷经碱性氧化得到酵。

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# Buchner-Curtius-Schlotterbeck 反应

### 羰基化合物和脂肪族重氯化物反应给出同系化的酮。

$$R^1$$
  $R^2$   $N_2$   $R^2$   $R^1$   $R^2$   $R^3$   $R^4$   $R^2$   $R^4$   $R^4$ 

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# Buchwald-Hartwig C-N 键和 C-O 键生成反应

芳基卤和胺在等摩尔量碱存在下直接由钯催化生成C-N键和C-O键:

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# Burgess 脱水剂

在 Ei 机理(发生消除反应时两个基团离去同时新键也随之形成)下 Burgess 脱水剂 是将仲醇或叔醇产生烯烃的有效试剂。

$$\begin{array}{c} \text{CH}_{3}\text{O}_{2}\text{C}-\bar{\textbf{N}}-\ddot{\ddot{\textbf{S}}}-\bar{\textbf{N}}\text{Et}_{3} \\ \text{R}^{1} & \text{CH}_{3}\text{O}_{2}\text{C}-\bar{\textbf{N}}-\ddot{\ddot{\textbf{S}}}-\bar{\textbf{N}}\text{Et}_{3} \\ \text{Burgess } 斌和 \\ \\ \text{R}^{2} & \text{CH}_{3}\text{O}_{2}\text{C}-\bar{\textbf{N}}-\ddot{\ddot{\textbf{S}}}-\bar{\textbf{C}}\text{I} \\ \\ \text{CH}_{3}\text{O}_{2}\text{C}-\bar{\textbf{N}}-\ddot{\ddot{\textbf{S}}}-\bar{\textbf{C}}\text{I} \\ \\ \text{CH}_{3}\text{O}_{2}\text{C}-\bar{\textbf{N}}-\ddot{\ddot{\textbf{S}}}-\bar{\textbf{N}}\text{Et}_{3} \\ \\ \text{CH}_{3} & \text{C}\text{C}\text{C}+\bar{\textbf{N}}-\ddot{\ddot{\textbf{N}}}-\bar{\textbf{N}}\text{Et}_{3} \\ \\ \text{CH}_{3} & \text{C}\text{C}+\bar{\textbf{N}}-\ddot{\ddot{\textbf{N}}}-\bar{\textbf{N}}\text{Et}_{3} \\ \\ \text{CH}_{3} & \text{C}\text{C}+\bar{\textbf{N}}-\ddot{\ddot{\textbf{N}}}-\bar{\textbf{N}}\text{Et}_{3} \\ \\ \text{CH}_{3} & \text{C}\text{C}+\bar{\textbf{N}}-\ddot{\ddot{\textbf{N}}}-\bar{\textbf{N}}\text{Et}_{3} \\ \\ \text{C}\text{C}\text{H}_{3} & \text{C}\text{C}+\bar{\textbf{N}}-\bar$$

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用碱在芳香醛、甲醛或其它无 $\alpha$ -氢的脂肪氢之间发生氧化还原反应给出相应的 醇和酸。

途径: a:

羧酸最后一步去质子化推动反应进行。

### 途径 b:

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# Kuru chenich Artentia

## Carroll 重排

碱催化下经负离子促进的 Claisen 反应将烯丙基醇和 8-酮酯转化为 Y-酮羰基 烯烃。

$$R^1$$
  $R^1$   $R^2$   $R^2$   $R^2$   $R^2$   $R^3$   $R^4$   $R^2$   $R^4$   $R^4$ 

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# Castro-Stephens 偶联(反应)

芳基炔合成、参见 Cadiot-Chodkiewicz 偶联(反应)。

$$Ar - X + Cu - R - R$$

$$ref. \qquad Ar - X + L_3Cu - R$$

$$Ar - X + L_3Cu - R$$

$$Ar \times Cu - R$$

$$CuX + Ar - R$$

另一个过程和 Cadiot-Chodkiewicz 偶联相似:

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# North Electricity Artifaction of the State o

# Chapman 重排

### O(氧)-芳基亚胺基醚中的芳基热重排为酰胺

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## Chichibabin 胺基化反应

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# Chichibabin 吡啶合成

醛和氨缩合生成吡啶。

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### 黄原酸酯热消除为烯

$$R \xrightarrow{O \setminus S} \xrightarrow{\Delta} \xrightarrow{R \setminus O} \xrightarrow{Ei} S \xrightarrow{S}$$

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## Ciamician-Dennsted 重排

吡啶和从CHCI。和 NaOH 反应所产生的二氯卡宾发生环丙烷化,接着重排生成3-氯吡啶。

$$\begin{array}{c} \text{CHCl}_3 \\ \text{NaOH} \\ \text{CI}_2 = :CCl_2 \\ \text{卡宾} \\ \text{卡宾} \\ \text{HO} \\ \text{NaOH} \\ \text{N$$

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# Kuru Ciring Kari Kuru Chaisen 缩合(反应)

酯在碱催化下缩合为β-酮酯。

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# Claisen, Eschenmoser-Claisen, Johnson-Claisen, 利 Ireland-Claisen 重排

Claisen, Eschenmoser-Claisen, Johnson-Claisen, Ireland-Claisen, Para-Claisen 和 Carroll 重排都 周于[3, 3]—σ重排的同一类协同效应。此处箭头所示的仅仅是说明。异常的 Claisen 重排参见第一页。

### Claisen 重排

### Eschenmoser-Claisen (酰胺缩醛) 重排

Johnson-Claisen(原酸酯)重排

# kwk<sup>n</sup> Krand-Claisen (硅基酯)重排

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### Clark-Eschweiler 胺的还原烷基化

伯胺或仲胺用甲醛和甲酸进行还原甲基化反应。

$$\longrightarrow O=C=O\uparrow + H H \rightarrow R-N + R-N + R-N$$

$$- R \stackrel{+}{\longrightarrow} H \stackrel{-}{\longrightarrow} O \longrightarrow O = C = O \uparrow + R \stackrel{+}{\longrightarrow} H \stackrel{-}{\longrightarrow} R - N$$

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# Clemmensen 还原(反应)

用锌汞齐和氯化氢将醛和酮还原为相应的亚甲基化合物。

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# Combes 喹啉合成

酸催化下苯胺和β-二酮缩合形成喹啉。

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# Conrad-Lipach 反应

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## Cook-Heilbron 噻唑合成

从α-胺基腈和二硫化碳或氧硫化碳,并硫氰酸酯,二硫化酸合成5-氨基噻唑

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# Cope, 含氧的 Cope 和负离子含氧 Cope 重排

Cope, 含氧 Cope, 和负离子含氧 Cope 重排都属于  $\sigma$  重排 一类协同反应, 此处箭头所示仅仅是一个说明。

### Cope 重排

### 含氧的Cope重排

### 负离子含 Cope 重排

OH 
$$R$$
  $A$ ,  $[3,3]$ - $\sigma$   $\mathbb{I}$   $\mathbb{I}$ 

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# Corey-Bakshi-Shibata (CBS) 还原(反应)

酮在手性 噁 唑硼 烷催化下的对映选择性还原。

RANGE CHERTICAL AND SELECTION OF THE PROPERTY OF THE PROPERTY

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## Corey-Chaykovsky 反应

Corey-Chaykovsky反应指疏叶立德,如二甲基亚甲基硫氧1(有时又称Corey叶立德,用DMSY表示)、或二甲基硫亚甲基2与亲申物种3,如羰基、烯烃、亚胺、或硫羰基反应生成如4那样相应的环氧化物,环丙烷,氮杂环丙烷或硫杂环丙烷。

X = O,  $CH_2$ ,  $NR^2$ , S,  $CHCOR^3$ ,  $CHCO_2R^3$ ,  $CHCONR_2$ , CHCN

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# Corey-Fuchs 反应

醛发生一碳同系化生成二溴烯烃,然后用 BuLi 处理生成终端炔烃。

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## Corey~Kim 氧化(反应)

醇在 NCS/DMF 作用后,再用碱处理可氧化为相应的醛或酮。

OH 1) NCS, DMS
$$R^{1} \stackrel{(R^{2})}{R^{2}} \stackrel{(CH_{3})_{2}}{R^{1}} \stackrel{(CH_{3})_{2}}{R^{2}}$$

$$R^{1} \stackrel{(CH_{3})_{2}}{R^{2}} \stackrel{(CH_{3})_{2}}{R^{2}} \stackrel{(CH_{3})_{2}}{R^{2}}$$

或者:

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# Corey-Winter olefin 烯烃合成

邻二醇经1,1-硫代羰基二咪唑和三甲氧基膦处理转化为相应的烯。

# www.itetijer aktikuti.

### 经热解实验研究表明反应过程卡宾中间体存在:

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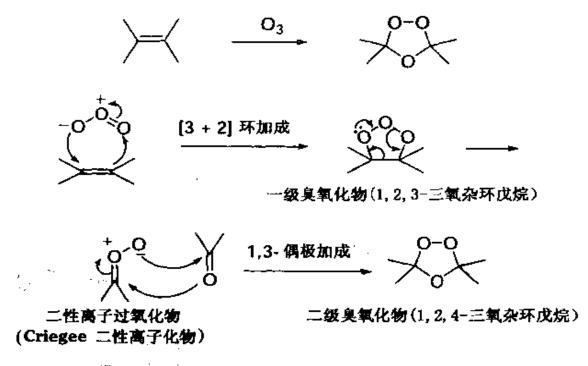
WW. Steeligt A. 48

## Criegee 邻二醇裂解(反应)

邻二醇由 Pb(OAc)4 氧化为相应的两个羰基化合物。

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## Curtius 重排

### 酰基叠氮化物径异氰酸酯中间体发生热分解。

$$\begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \\ \\ \end{array} \end{array} \end{array} \end{array} \begin{array}{c} \begin{array}{c} \begin{array}{c} \\ \end{array} \end{array} \end{array} \begin{array}{c} \begin{array}{c} \\ \\ \end{array} \end{array} \begin{array}{c} \\ \end{array} \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \end{array} \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \end{array} \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \end{array} \end{array} \begin{array}{c} \\ \end{array} \\ \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c}$$

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## Dakin-West 反应

α-氨基酸酰基化后径氮杂内酯中间体生成α 酰胺酮。

$$R \downarrow CO_2H$$
  $AC_2O$   $R \downarrow NHAC$   $NHAC$   $R \downarrow CO_2H$   $R \downarrow OH$   $NH_2$   $NH_$ 

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 $\alpha$ , $\beta$  - 不饱和酮和三甲基硅基内二烯在 Lewis 酸存在下进行的生成 三甲基硅基环戊基的成环反应。

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## Darzens 缩水甘油酸酯缩合(反应)

碱催化下从 $\alpha$ -卤代酯和羰基化合物生成 $\alpha$ , $\beta$ -环氧酯(缩水甘油酯)。

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CO<sub>2</sub>Me

# 

## de Mayo 反应

从1,3-二酮产生的烯醇与烯烃经[2+2]光加成接着进行逆 Aldol 反应给出1,5-二

较少的位置异构体:

CO<sub>2</sub>Me

CO₂Me

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# Kuru dienida, Arthatilda

# Demjanov 重排

伯胺经重氮化后发生碳正离子重排为醇。

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# Dess-Martin 过碘酸酯氧化(反应)

AcO OAC
$$R^{1} R^{2}$$

$$R^{1} R^{2}$$

$$R^{1} R^{2}$$

$$R^{1} R^{2}$$

$$R^{2} R^{1} R^{2}$$

$$R^{1} R^{2}$$

$$R^{1} R^{2}$$

$$R^{2} R^{1} R^{2}$$

$$R^{1} R^{2}$$

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# Dieckmann 缩合(反应)

Dieckmann 缩合是 Claisen 缩合的分子内过程。

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### kuru che Rugue che Rugue

Diels-Alder 反应; 反转电子要求的 Diels-Alder 反应,杂原子参与的 Diels-Alder 反应, 杂原子参与的 Diels-Alder 反应, 都属于[4+2]一类协同过程的环加成反应。此处箭头所示 的仅仅是说明。

### 正常的 Diels-Alder 反应

如:

如:

$$CO_2Me$$
 $\Delta$ 
 $CO_2Me$ 

### 杂原子参与的 Diels-Alder 反应

### 杂原子二烯与亲双烯加城

$$\begin{array}{c|c} SO_2Ph \\ EtO_2C \\ N \\ Ph \end{array} \begin{array}{c} OEt \\ Ph \\ OEt \\ Ph \end{array} \begin{array}{c} SO_2Ph \\ EtO_2C \\ N \\ OEt \\ Ph \end{array}$$

### b. 二烯和杂原子亲双烯体加成

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# Marking Marking Dienone-phenol rearrangement 二烯酮-酚重排

### 酸促进的4,4-二取代环己二烯酮重排为3,4-二取代酚。

$$\begin{array}{c} & & & \\ & &$$

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### 二-π- 甲烷重排

### 1,4-二烯光解生成乙烯基环丙烷

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# Doebner-

## Doebner-von Miller 反应

Doebner-von 反应是 Skraup 喹啉合成法 (P. 378) 的另一种形式。因此,后者的机理也适用于前者。下面所示的另一种机理基于这样一个事实,即事先(另外) 加入亚胺(Schiff碱)也给出 2-甲基喹啉。

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# Doering-LaFlamme 丙二烯合成

烯烃用溴仿及烷氧化物处理后生成的同碳二溴环丙烷再反应生成丙二烯。

$$R^1$$
  $R^3$   $CHBr_3$   $R^1$   $R^2$   $R^3$   $R^4$   $R^4$ 

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芳香醛和 α-硝基酯缩合生成异 哪唑。有意思的是,异 哪唑环上的氯原子是来自于硝基的。

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### Dötz 反应

从烯基烷氧基铬卡宾化物 (Fischer 卡宾) 和炔烃生成 Cr(CO)。配位的氢醌。

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# Road 扩环反应

### 2-卤代甲基环酮径自由基参与的扩环反应

由相邻的酚羰基所产生在这个裂分模式中的环丙基氧自由基中间体碎片而生成的 叔自由基因共振效应得到稳定。

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磺酰胺和从胺而生成的重氮盐反应生成叠氮化物。

$$Ar \xrightarrow{NH_2} \xrightarrow{HNO_2} \xrightarrow{Ar} \xrightarrow{N_2^+} \xrightarrow{1. TsNH_2} \xrightarrow{N_3}$$

$$1. TsNH_2 \xrightarrow{2. TOH} \xrightarrow{N_3}$$

$$2. TOH \xrightarrow{N_1} \xrightarrow{N_2} \xrightarrow{N_2} \xrightarrow{N_2} \xrightarrow{N_3} \xrightarrow{N_2} \xrightarrow{N_2} \xrightarrow{N_2} \xrightarrow{N_2} \xrightarrow{N_2} \xrightarrow{N_3} \xrightarrow{N_1} \xrightarrow{N_2} \xrightarrow{N_2$$

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终端炔烃在化学计量(常常过量)Cu(OAc)2促进下发生的氧化偶联反应。另一类 Glaser偶联反应见第160页。

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## Kurulifa (18) Eschenmoser 偶联反应

如:

$$t \cdot BuO_2C \longrightarrow S \xrightarrow{1. BrCH_2CO_2CH_3} t \cdot BuO_2C \longrightarrow Bn$$

$$t \cdot BuO_2C \longrightarrow S \xrightarrow{2. PPh_3, Et_3N, 90\%} t \cdot BuO_2C \longrightarrow S \xrightarrow{EBuO_2C} CO_2CH_3$$

$$t \cdot BuO_2C \longrightarrow S \xrightarrow{EBuO_2C} CO_2CH_3 \xrightarrow{EBuO_2C} CO_2CH_3$$

$$t \cdot BuO_2C \longrightarrow S \xrightarrow{EBuO_2C} CO_2CH_3 \xrightarrow{EBuO_2C} CO_2CH_3$$

$$t \cdot BuO_2C \longrightarrow S \xrightarrow{EBuO_2C} CO_2CH_3 \xrightarrow{EBuO_2C} CO_2CH_3$$

$$t \cdot BuO_2C \longrightarrow S \xrightarrow{EBuO_2C} CO_2CH_3 \xrightarrow{EBuO_2C} CO_2CH_3$$

$$t \cdot BuO_2C \longrightarrow S \xrightarrow{EBuO_2C} CO_2CH_3 \xrightarrow{EBuO_2C} CO_2CH_3$$

$$t \cdot BuO_2C \longrightarrow S \xrightarrow{EBuO_2C} CO_2CH_3 \xrightarrow{EBuO_2C} CO_2CH_3$$

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### 用铬酰氯氧化芳(基)甲基为相应的芳基醛。

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KANTA CHERTICAL AND SELECTION OF THE SERVER SERVER

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### Favorskii 重排

### Quasi-Favorskii 重排

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AND THE PROPERTY OF THE PROPER

### Feist-Bénary 呋喃合成

 $\alpha$ -卤代酮和 $\beta$ -酮酯在吡啶存在下反应生成呋喃。

$$CO_2Et$$
 $CO_2Et$ 
 $CO_2Et$ 

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由于异头效应而有利于a 键加成。

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N-亚硝基苯胺转化为对亚硝基苯胺。参见 Orton 重排。

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## Fischer-Speier 酯化(反应)

又常称为 Fischer 酯化反应, 是醇和羧酸在质子酸催化下的酸化反应。

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### Forster 反应

从 α-羟亚氨基(肟基)酮生成 α-重氨酮。

### 或者:

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## Fritsch-Buttenberg-Wiechell 重排

1,1-二芳基-2-卤代乙烯用碱处理经亚烷基卡宾中间体转变为二芳基乙炔。

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### Fujimoto-Belleau 反应

烯醇内酯与格民试剂反应生成环状 $\alpha$ -取代- $\alpha$ , $\beta$ -不饱和酮。是对 Robinson 环化反应的另一种选择。

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### WWW. Cheerilich Arthur Arthur

### Fukuyama 胺合成

### 用2,4-二硝基苯磺酰氯将伯胺转变为仲胺。

$$R^{1}NH_{2}$$
 $NO_{2}$ 
 $NO_{2}$ 

### 机理请见Mitsunobu反应(p. 265)。

Meisenheimer 络合物

$$\longrightarrow \begin{array}{c} H \\ R^{1} \stackrel{N}{\longrightarrow} R^{2} \end{array} + \begin{array}{c} S \stackrel{CO_{2}H}{\longrightarrow} NO_{2} \\ NO_{2} \end{array} + SO_{2}$$

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### 

CO和HCI、AICI。高压下在芳环发生的甲酰化反应。

$$+ CO + HCI \xrightarrow{AICI_3} CHO$$

$$: \overline{C} = \overline{O}: \qquad : C = \overline{O} \xrightarrow{AICI_3} : C = \overline{O} \xrightarrow{AICI_3}$$

$$+ CI \xrightarrow{CI} : C = \overline{O} \xrightarrow{AICI_3} : C = \overline{O} \xrightarrow{AICI_3}$$

$$+ CI \xrightarrow{AICI_3} : C = \overline{O} \xrightarrow{AICI_3} \xrightarrow{CI} \xrightarrow{AICI_3}$$

$$+ CI \xrightarrow{AICI_3} : C = \overline{O} \xrightarrow{AICI_3} \xrightarrow{CI} \xrightarrow{AICI_3}$$

$$+ CI \xrightarrow{AICI_3} : C = \overline{O} \xrightarrow{AICI_3} \xrightarrow{CI} \xrightarrow{AICI_3}$$

$$+ CI \xrightarrow{AICI_3} : C = \overline{O} \xrightarrow{AICI_3} \xrightarrow{CI} \xrightarrow{AICI_3}$$

$$+ CI \xrightarrow{AICI_3} : C = \overline{O} \xrightarrow{AICI_3} \xrightarrow{CI} \xrightarrow{AICI_4} \xrightarrow{CI} \xrightarrow{AICI_4} \xrightarrow{CHO} \xrightarrow{CHO} \xrightarrow{AICI_4} \xrightarrow{CHO} \xrightarrow{AICI_4} \xrightarrow{CHO} \xrightarrow{AICI_4} \xrightarrow{CHO} \xrightarrow{AICI_4} \xrightarrow{CHO} \xrightarrow{CHO} \xrightarrow{AICI_4} \xrightarrow{CHO} \xrightarrow{CHO} \xrightarrow{AICI_4} \xrightarrow{CHO} \xrightarrow$$

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### Gewald 胺基噻吩合成

酮、 α-活化亚甲基腈和元素磷在碱促进下生成胺基噻吩。

$$S_7 + R CO_2R^2$$

$$R CO_2R^2$$

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碱促进下芳基重氮盐和一个芳烃之间径自由基偶联生成一个二芳基化合物。

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### Gribble 吲哚还原

用NaBHaCN在冰醋酸中还原吲哚双键,而用NaBHa则可导致还原和N-烷基化。

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# Kuru Ling (1887) Gribble 二芳基酮还原

用 NaBH<sub>4</sub> 在三氟乙酸中还原二芳基酮和二芳基甲醇为二芳基甲烷,也可应用于二杂芳环酮和醇的还原。

$$Ar^{1} Ar^{2} \xrightarrow{NaBH_{4}} Ar^{1} Ar^{2} \xrightarrow{Ar^{1} Ar^{2}} Ar^{1} Ar^{2}$$

$$Ar^{1} Ar^{2} \xrightarrow{Ar^{2}} Ar^{2} \xrightarrow{Ar^{1} Ar^{2}} Ar^{2}$$

$$(CF_{3}CO_{2})_{3}B-H$$

$$Ar^{1} Ar^{2} \xrightarrow{HB(OCOCF_{3})_{3}} Ar_{1} Ar^{2}$$

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# Grignard 反应

从有机卤化物和金属镁产生的有机镁化物(格氏试剂)对亲电物种的加成反应。

$$R-X \xrightarrow{Mg(0)} R-MgX \xrightarrow{R^1} R^2 \xrightarrow{R^1} R^2$$

#### 格氏试剂的生成:

#### 格氏反应: 离子机理、

自由基机理,

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## Grob 碎裂化

通式:

$$X = OH_2^+$$
, OTs, I, Br, Cl;  $Y = O^-$ ,  $NR_2$ 

例:

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# Guareschi-Thorpe 缩合

#### 氰基乙酸乙酯和乙酰乙酸在氨存在下生成 2-吡啶酮。

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# Hajos-Wiechert 反应

(S)-(-)- 脯氨酸催化的不对称 Robinson 环化反应。

ANN STREET, CT. 18 HE HE SEE STREET, CT. 170

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# kwky Cur 成於 Haller-Bauer 反应

不能烯醇化的酮在碱引发下生成羧酸衍生物和一个羰基被氢所取代的中性碎片。

$$R^{1}$$
  $R^{2}$   $R^{3}$   $R^{4}$   $R^{4}$   $R^{5}$   $R^{1}$   $R^{2}$   $R^{5}$   $R^{4}$   $R^{5}$   $R^{1}$   $R^{2}$   $R^{5}$   $R^{5}$   $R^{1}$   $R^{2}$   $R^{5}$   $R^{5}$   $R^{1}$   $R^{2}$   $R^{5}$   $R^{5}$   $R^{1}$   $R^{2}$   $R^{3}$   $R^{4}$   $R^{5}$   $R^{1}$   $R^{2}$   $R^{3}$   $R^{4}$   $R^{5}$   $R^{1}$   $R^{2}$   $R^{3}$   $R^{4}$   $R^{5}$   $R^{5}$   $R^{1}$   $R^{2}$   $R^{3}$   $R^{4}$   $R^{5}$   $R^{5}$   $R^{1}$   $R^{2}$   $R^{3}$   $R^{4}$   $R^{5}$   $R^{$ 

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### Hantzsch 吡啶合成

醛,β-酮酯和氨缩合生成二氢吡啶。

$$CO_2$$
Et  $NH_3$   $EtO_2$ C  $R$   $CO_2$ Et  $R_1$   $R_2$   $R_3$   $R_4$   $R_4$   $R_5$   $R_4$   $R_5$   $R$ 

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### Hantzsch 吡咯合成

 $\alpha$  - 氯甲基酮和  $\beta$  - 酮酯及氨组合成吡咯的反应。

$$CO_2$$
Et  $NH_3$   $NH_3$   $CO_2$ Et  $NH_3$   $NH_4$   $NH_5$   $NH$ 

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# Raworth 反应

芳环和丁二酐发生 Friedl-Crafts 反应接着还原和再一次分子内的 Friedl-Crafts 反应给出四氢萘酮。这整个过程称为 Haworth 反应。

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邻-苯甲酰基苯甲酸在 H<sub>2</sub>SO<sub>4</sub> 或 P<sub>2</sub>O<sub>5</sub> 存在下的重排。

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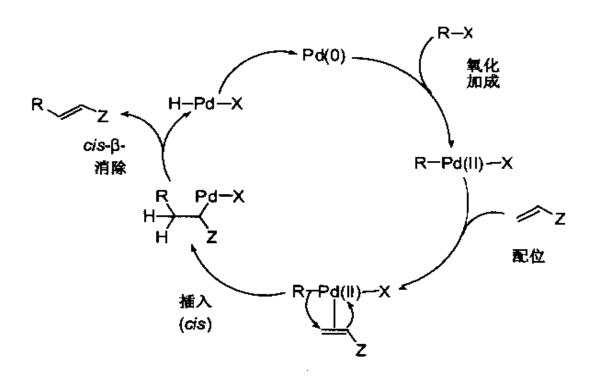
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### Heck 反应

钯催化的有机卤代物或三氟磺酸酯和烯烃之间的偶联反应。

$$R-X$$
  $Pd(0)$   $R$   $Z$ 

X = I, Br, OTf, etc.  $Z = H, R, Ar, CN, CO_2R, OR, OAc, NHAc, etc.$ 



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# Kurwichenich Artstellen

# Hegedus 吲哚合成

化学计量 Pd(II) 促进的烯基苯胺氧化环化为吲哚。参见 Wacker 氧化。

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## Hell-Volhardt-Zelinsky 反应

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# <sub>Kuru</sub>cri<sup>(K)</sup> Henry (硝醇 ) 反应

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## Herz 反应

由苯胺用单氯化硫处理生成噻唑硫鉛氯化物,再用碱处理后给出α-胺基硫酚。

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发生在杂芳基受体上的分子内或分子间的Heck反应。

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# Hiyama 交叉偶联反应

钯催化有机硅和有机卤代物或三氟磺酸酯等在诸如F<sup>-</sup>、OH<sup>-</sup>之类活化剂存在下发生的交叉俱联反应(无活化剂存在时金属的转移较难进行)。催化循环参见P. 234 的 Kumada 偶联。

$$Pd(Ph_3P)_4$$
,  $TBAF$ 
 $Pd(Ph_3P)_4$ ,  $TBAF$ 
 $Pd(Ph_3P)_4$ ,  $TBAF$ 
 $Pd(Ph_3P)_4$ ,  $TBAF$ 
 $Pd(Ph_3P)_4$ 
 $Pd(Ph_3P)_4$ 

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# Hoch-Campbell 氮杂环丙烷合成

酮肟和过量的格氏试剂反应后形成的有机金属络合物水解生成氮杂环丙烷。

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# WANT HEALTH HERETHAN

# Hodges-Vedejs 噁唑 金属化作用

**噖唑金属化后与苯甲醛作用生成主要在4-位取代的 噖 唑 产物□:** 

#### 2-暖隆锂

然而,开环过程可**被硼烷排**止<sup>[3]</sup>:

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## Hofmann 重排 (Hofmann 降解反应)

伯酰胺用次卤酸处理,经异氰酸酯中间体莆成少一个碳原子的伯胺。

$$R = N = C = O$$
  $R = N = C = O$   $R = N = C$   $R = C$ 

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# Hofmann-Löffler-Freytag 反应

质子化的N-氯代胺经热取光化学分解开成四氢吡咯或哌啶。

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# Hofmann-Martius 反应

芳基烷基胺的氯化氢盐受热后发生烷基的迁移反应。

Reilly-Hickinbottom 重排是 Hofmann-Martius 反应的扩展,反应用 Lewis 酸替代质子酸,反应途径基本相似。

$$ZnCl_2$$
  $+$   $NH_2$   $NH_2$   $+$   $NH_2$ 

# 

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# Hooker 氧化

2-羟基-3-烷基-1,4-醌和 KMnO, 氧化导致侧链烷基失去一个亚甲基, 同时羟基 和烷基的位置互变。

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### Horner-Wadsworth-Emmons 反应

从醛和膦酸酯生成烯烃。副产物为水溶性磷酸盐,故后处理较相应的 Wittig 反应简单得多。

赤式(动力学)或苏式(热力学)

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# Ing-Manske 过程

用肼从邻苯二甲酰亚胺中放出相应胺的一类 Gabriel 胺的合成。

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# Jacobsen-Katsuki 环氧化

Mn(III) 催化的(2)-烯烃的不对称环氧化

#### 1. 协同的氧转移 (cis-环氧化物):

$$R = R_1 + O \longrightarrow R_1 \longrightarrow R_$$

#### 2. 经自由基中间体 (trans-环氧化物) 的氧转移:

$$R = R_1 + 0 \\ M_{IN}(V) \longrightarrow R_0 \\ M_{IN}(V) \longrightarrow R_1 \\ M_{IN}(V) \longrightarrow R_$$

#### 3. 经 Mn-O杂环中间体 (cis-环氧化物) 的氧转移;

$$\frac{R}{R_1 + Mn(V)} \xrightarrow{R_1 + Mn(V)} \frac{R}{R} \xrightarrow{R} \frac{R} \frac{R}{R} \xrightarrow{R} \frac{R}{R} \xrightarrow{R} \frac{R}{R} \xrightarrow{R} \frac{R}{R} \xrightarrow{R} \frac{R}{R} \xrightarrow{$$

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ANN STEEL ST

# Jacobsen 重排

多烷基或多卤代苯用浓硫酸处理给出重排物的磺酸化的多烷基或多卤代苯。

机理 1:

机理 2:

# Name Chestical At the State of the State of

#### 机理 3:

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# Japp-Klingemann 腙合成

α-酮酯和重氮盐在碱促进下生成腙。

$$ArN_2^+ Cl^- + CO_2R$$
  $CO_2R$   $ROH$   $RO$ 

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# Julia-Lythgoe 成烯(反应)

从砜和醛生成(E)-烯烃

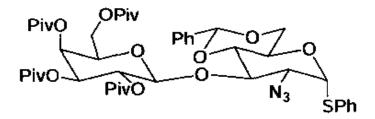
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# Kahne 苷化作用

在三氟甲基 磺酸酐活化下,异头中心上的亚砜作为苷化受体发生非对映选择性苷化 效应。



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### Keck 立体选择性烯丙基化(反应)

醛和烯丙基锡在一个Lewis 酸和催化量的手性BINAP(或其它手性配体)存在下发生不对称的烯丙基化反应。

对映选择性是通过取代掉 $Ti(OPr^i)_4$ 上异丙氧基的手性配体所具有的立体轴得以实现的。手性Lewis酸成为 $Ti(OPr^i)_2$ (BINAP),它实际上是不稳定的。

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# Keck 大环内酯化(反应)

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# Kemp 消除(反应)

苯并 噁 唑用碱处理后开环生成水杨腈。

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# Kennedy 氧化周环反应

δ-羟基烯与 Re<sub>2</sub>O<sub>7</sub> 作用不对称合成四氢呋喃。

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# Kharasch 加成反应

过渡金属催化的 CXCI3 对烯烃的自由基加成。

$$R^{2}$$
 +  $CXCI_{3}$   $M$   $R^{1}$   $CI$   $X = H, CI, Br$ 

M: Ru, Re, Mo, W, Fe, Al, B, Cr, Sm, 等金屬试剂

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# Knoevenagel 缩合(反应)

羰基化合物和活化的亚甲基化合物在胺催化下的缩合。

ANT A 大献

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### Knorr 吡咯合成

Feist-Bénary呋喃合成法 (P. 134)的改进。从 $\alpha$ -卤代酮氨化后生成的 $\alpha$ -氨基 酮和β-酮酯反应生成吡咯。

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# Koch 羰基化反应 (Koch-Haaf 羰基化反应)

强酸催化的从醇或烯烃和 CO 生成叔烷基羧酸的反应。

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# Koenig-Knorr 苷化(反应)

从  $\alpha$ - 卤代糖在银盐的影响下产生  $\beta$ - 苷。

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### Kolbe 电解偶联

羧酸盐电解生成偶联物。两个羧酸盐相同的二聚后形成的产物是对称的,两个相同的酸羧酸盐反应后生成不对称的产物。

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# Kolbe-Schmitt 反应

酚氧钠和CO。发生羧基化反应,主要在邻位。

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### Kostanecki 反应

也称 Kostanecki-Robinson 反应。  $1\rightarrow 2$  的反应代表了一个 Allan-Robinson 反应(见p. 3),而  $1\rightarrow 3$  是一个 Kostanecki(酰基化)反应。

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# Krapcho 脱羰反应

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# Kröhnke 反应(吡啶合成)

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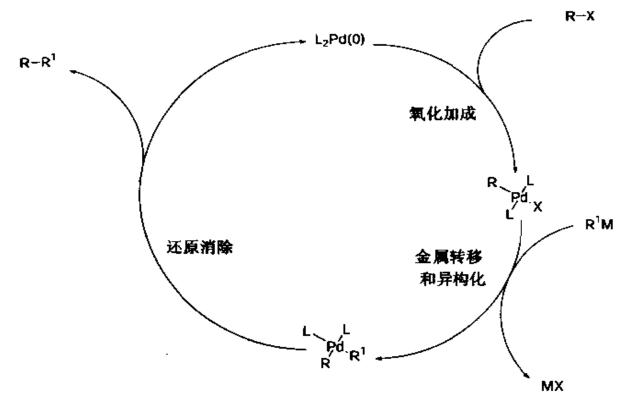
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### Kumada 交叉偶联反应

Kumada 交叉偶联反应(有时也称Kharasch交叉偶联反应)是在 Ni 或 Pd催化下在格氏试剂和一个有机卤代物或三氟磺酸酯等之间进行的交叉偶联反应。

Kumada 交叉偶联反应 和 Negishi、Stille、Hiyama 及 Suzuki 等人名反应一样,都属于同一类 Pd 催化的有机卤代物或有机 三氯磺酸酯或其它亲电物种与有机 金属试剂之间的交叉偶联反应。这些反应都有如下页所示的催化循环。但 Hiyama 反应和 Suzuki 反应稍有不同,有额外的活化步骤来实现金属转移作用。

催化循环为:



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### Larock 吲哚合成

邻碘苯胺和一个丙炔醇在 Pd 催化下发生偶联反应生成吲哚。

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# Lawesson (硫羰基化) 试剂

2,4-双-(4-甲氧基苯基)-1,3-二硫-2,4-二膦杂环丁烷将酮、酰胺和酯中的羰基转变为相应的硫羰基化合物。

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# kwrwitting Leuckart-Wallach 反应

从羰基化合物和一个胺在过量甲酸存在下发生还原胺基化来合成胺,甲酸在这里 提供氧而起到还原剂作用。

$$\stackrel{R^1}{\underset{R^2}{\longleftarrow}} O + HN \stackrel{R^3}{\underset{R^4}{\longleftarrow}} \frac{HCO_2H}{\underset{R^2}{\longleftarrow}} \stackrel{R^1}{\underset{R^4}{\longleftarrow}} \stackrel{R^3}{\underset{R^2}{\longleftarrow}} + CO_2\uparrow + H_2O$$

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# Liebeskind-Srogl 偶联(反应)

在硫醇酯和有机硼酸之间经 Pd催化发生交叉偶联生成酮。

R S R Pd(0)L<sub>2</sub>

R Pd(0)L<sub>2</sub>

$$R^1$$
 $R^2$ 
 $R^2$ 
 $R^2$ 
 $R^2$ 
 $R^2$ 
 $R^3$ 
 $R^4$ 
 $R^2$ 
 $R^2$ 
 $R^3$ 
 $R^4$ 
 $R^2$ 
 $R^4$ 
 $R^2$ 
 $R^4$ 
 $R^4$ 

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### Lossen 重排

#### O(氧)-酰基化的羟肟酸用碱处理生成异氰酸酯。

$$R^{1}$$
  $N$   $R^{2}$   $O$   $R^{1}$   $N$   $C$   $R^{1}$   $N$   $C$   $R^{2}$   $R^{1}$   $N$   $C$   $R^{2}$   $R^{$ 

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### McMurry 偶联(反应)

羰基用低价钛,如从TiCl<sub>3</sub>/LiAlH<sub>4</sub>产生的Ti(0)处理得到烯基,反应是一个单电子过程。

$$R^1$$
  $R^1$   $R^1$   $R^1$   $R^1$   $R^1$   $R^2$   $R^2$ 

氧化物覆盖的钛表面

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# Mannich 反应

胺、甲醛和一个带有酸性亚甲基成分的化合物之间的三组分反应发生胺甲基化 反应。

R = H, \*NH2=CH2 的盐又称为 Eschenmoser 盐

$$R^{1} \xrightarrow{Q} R^{2} \xrightarrow{\text{H}^{+}} R^{1} \xrightarrow{R^{+}} R^{2} \xrightarrow{R^{-}} R^{1}$$

Mannich 反应也可在碱性条件下发生:

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参见 Grob 碎片化反应 (P. 166)。事实上, Marshall 硼化物碎片化也属于 Grob 碎片化一类。

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### Martin 硫化物脱水剂

仲醇和叔醇脱水成烯,但伯醇生成醚。参见 Burgess 脱水剂。

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# Masamune-Roush 条件

用于对碱敏感的醛和膦酸酯的 Horner-Wadsworth-Emmons 反应。

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# Meinwald 重排

双环环氧化物经酸处理生成重排的醛。

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#### Meisenheimer 络合物

也称为 Meisenheimer-Jackson 盐, 是在某些 S<sub>N</sub>Ar 反应中出现的稳定的中间体。

Meisenheimer 络合物 ( Meisenheimer-Jackson 盐 )

用Sanger 试剂的反应比用相应的氯、溴或碘代二硝基苯的反应要快一点,因为氯的吸电子能力量强。反应速度与高去基团的高去能力无关。

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[1,2]- σ重排:

[2,3]- σ重排:

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# Meyer-Schuster 重排

 $\alpha$ -炔基取代的仲或叔醇经1,3-迁移异构化为 $\alpha$ , $\beta$ -不饱和羰基化合物。 当炔基是终端炔时,产物是醛,炔基为中间炔时,产物为酮。

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### Michael 加成(反应)

亲核的碳原子对α,β-不饱和体系的共轭加成。

例:

例:

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从卤代烃和亚磷酸酯生成膦酸酯。

通式:

$$(R^{1}O)_{3}P + R_{2}-X \xrightarrow{\Delta} R_{2}-P-OR^{1} + R^{1}-X$$

 $R^1$ = 烷基,等;  $R_2$ = 烷基,酰基等; X = Cl, Br, I

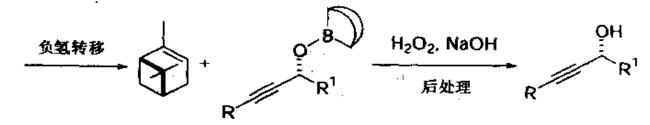
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# Midland 还原(反应)

用 B-3-α-蒎烯-9-BBN 对酮进行不对称还原。

$$\begin{bmatrix}
A & B & O \\
A & B & O \\
A & B & C
\end{bmatrix}$$



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# Miller-Snyder 芳氰合成

先后用羟胺及对硝基苯腈将醛转变为腈。

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# Morgan-Walls 反应 (Pictet-Hubert 反应)

#### Morgan-Walls 反应

#### Pictet-Hubert 反应

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# Mori-Ban indole 吲哚合成

邻卤代苯胺和侧链烯烃之间的分子内Heck 反应制备吲哚。

Pd(OAc)2 还原为 Pd(0):

Mori-Ban 吲哚合成:

Åc

the Property of All Age (2012) and

Pd(0) 再生:

$$H-PdBrL_n + NaHCO_3 \longrightarrow Pd(0) + NaBr + H_2O + CO_2\uparrow$$

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## Morin 重排

配尼西林砜经酸催化转变为头孢霉素,该重排可能对其它杂环类亚砜化合物 也是适用的。

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# Mukaiyama 醇醛反应

Lewis 酸催化的醛和硅基烯醇醚之间的 Aldol 缩合反应。

$$R-CHO + R^{1} \longrightarrow OSiMe_{3} \xrightarrow{Lewis *R} R^{OH} \cap R^{2}$$

### Mukaiyama Michael 加成

Lewis 酸催化的硅基烯醇醛对  $\alpha$  ,  $\beta$ -不饱和体系的 Michael 加成。

$$R^{2}$$
 Lewis  $R^{2}$   $R^{2}$   $R^{2}$   $R^{2}$   $R^{2}$   $R^{3}$   $R^{2}$   $R^{3}$   $R^{2}$   $R^{3}$   $R^{4}$   $R^{2}$   $R^{3}$   $R^{4}$   $R^{4}$ 

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## Mukaiyama 酯化(反应)

用类似2-氯-1-甲基吡啶碘化物(Mukaiyama 试剂)之类 Mukaiyama 试剂进行的 酯化反应。

通式:
$$R_{1}CO_{2}H + R_{2}OH \xrightarrow{R_{3}} (Mukaiyama 试剂) R_{1} O^{R_{2}} + O^{R_{3}}$$

$$X = F, Cl, Br$$
例:
$$GO_{2}H + HO \xrightarrow{HO-Bn} Br \xrightarrow{NAr} B$$

用 Mukaiyama 试剂生成酰胺的过程具有相似的机理 [4]。

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有时又称为 Schmittel 环化(反应)。 参见 Bergman 环化(反应)。

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### 酸催化下双乙烯酮进行电环化反应生成环戊烯酮。

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# Neber 重排

### 从对甲苯磺酰酮肟和碱生成α-胺基酮。

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# RATION Negishi 交叉偶联反应

Pd 催化的有机锌和有机卤代物,三氟磺酸酯等之间发生的交叉偶联反应。 催化循环可见 234页上的 Kumada 偶联反应。

$$R-X + R^{1}-ZnX$$
  $Pd(0)$   $R-R^{1} + ZnX_{2}$   $R-X + L_{2}Pd(0)$   $R-X + L_{2}Pd(0)$ 

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## Nenitzescu 吲哚合成

对苯酯和β-胺基丁烯酸酯缩合生成5-羟基吲哚。

中间的氧化-还原过程可能是双分子面对面的电子转移络合物(在硝基甲烷中)<sup>[3]</sup>:

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### Nicholas 反应

六羰基二钴化物稳定的炔丙基碳正离子被亲核物种捕获,接着氧化去金属化 给出炔丙基化合物。

**炔丙基正离子中间体(为六羰基二钴络合物所稳定)。** 

$$O=C=O\uparrow + \begin{pmatrix} (CO)_3Co-Co(CO)_2 \\ R^1 & & \\ \hline R^4 R^3 \end{pmatrix} \longrightarrow R^1 \longrightarrow \begin{pmatrix} Nu \\ R^4 \end{pmatrix}$$

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# Noyori 不对称氢化(反应)

羰基在 Ru([I]) BINAP 络合物催化下发生不对称氢化还原反应。

$$R^{OO} = \frac{H_2}{(R)-BINAP-Ru} = \frac{OH}{RPh_2}$$

$$R^{OH} = \frac{RPh_2}{RuCl_2L_2}$$

- HCI

催化循环:

$$\begin{array}{c} \text{Solv} & \text{(binap)CIHRu} \\ \text{(binap)CIHRu} & \text{OP} \\ \\ \text{[RuHCl(binap)(solv)_2]} & \text{(binap)CIHRu} \\ \text{H} & \text{R} \\ \\ \text{H}^+, \text{solv} & \text{OP}^1 \\ \\ \text{[RuCl(binap)(solv)_2]}^+ & \text{OH}^+ \text{OP}^1 \\ \\ \text{RuCl(binap)(solv)_2]}^+ & \text{OH}^+ \text{OP}^1 \\ \\ \text{OP}^1 & \text{OP}^1$$

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# Nozaki-Hiyama-Kishi 反应

Cr-Ni 双金属催化下的烯基卤对醛的氧化还原加成。

$$R$$
  $R^{1}$ — $CHO$   $R^{1}$   $R$ 

转移金属化并随后由二甲硫醛 (SMe2) 还原

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## Oppenauer 氧化

烷氧基催化的仲醇氧化反应。

OH 
$$R_1$$
  $R_2$  OH  $R_1$   $R_2$   $R_1$ 

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# Orton 重排

CI

## N-氯代酰基苯胺转变为相应的酰基对氯苯胺。参见Fischer-Hepp重排。

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## Overman 重排

丙烯醇经三 氯乙酰亚胺酯中间体立体选择性地转变为N-烯丙基 氯乙酰亚胺。

$$CCI_3$$
  $CCI_3$   $CCI$ 

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## Paal-Knorr 呋喃合成

酸催化下1,4-二酮环合成呋喃。

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# Paal-Knorr 吡咯合成

### 1,4-二酮和伯胺(或氨)给出吡咯。

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## Parham 环化(反应)

带有吸电子部份侧链的芳基卤经有机锂试剂处理,发生卤素-金属交换后接着亲核环化生成四元环或七元环。

$$\begin{array}{c} CH_3O \\ CH_3O \\ CH_3O \\ \end{array} \begin{array}{c} 2.2 \text{ eq. } \text{$\ell$-BuLi} \\ \end{array} \begin{array}{c} CH_3O \\ CH_3O \\ \end{array} \begin{array}{c} CH_3O \\ \end{array} \begin{array}{c} Et_2N \\ CH_3O \\ \end{array} \begin{array}{c} CH_3O \\ \end{array} \begin{array}{c} Et_2N \\ CH_3O \\ \end{array} \begin{array}{c} CH_3O \\ \end{array} \begin{array}{c}$$

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### Passerini 反应

羧酸、异腈和含氧化合物三组份缩合生成 α-酰基碳酰胺。参见 Ugi 反应。

$$R^{1}-\stackrel{+}{N}\equiv\stackrel{-}{C} + \stackrel{O}{R^{2}} + R^{4}-CO_{2}H \longrightarrow R^{1}-\stackrel{H}{N} + R^{2} R^{3} \stackrel{O}{O} + R_{4}$$

异腈

職基特移 
$$R^4$$
  $O$   $R^2$   $R^3$   $O$   $R^4$   $R^1$   $H^+$ 

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# <sub>kuru</sub>crifi<sup>ff,kx</sup>) Paterno-Büchi 反应

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NAMA CHENIER AND SOO

## Pauson-Khand 环戊烯酮合成

一个烯烃, 一个炔烃和CO在八羧基合二钴络合物促进下发生形式上[2+2+1]的环加成

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# Pechmann 缩合(反应)(香豆素合成)

Lewis 酸促进的酸和 $\beta$ -酮酯缩合生成香豆素。

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重氮化合物和炔烃发生1,3-偶极环加成反应生成吡唑。

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# Perkin

# Perkin 反应(肉桂酸合成)

芳香醛和乙酐反应合成肉桂酸。

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www.iterica ,标题提供用的

## Perkow 反应

从 α -卤代酮和亚膦酸三烷基酯合成膦酸烯醇酯。

通式:

X = Cl, Br, I, 适于仲或叔卤代物以抑止 Michaelis-Arbuzov 反应(见233页) 的发生

例:

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### Peterson 成烯反应

从 $\alpha$ -硅基碳负离子和羰基化合物生成烯烃。也称为含硅的 Wittig 反应。

### 碱性条件:

$$R^1$$
  $R^2$   $R^3$   $R^3$   $R^3$   $R^3$   $R^4$   $R^2$   $R^3$   $R^3$   $R^4$   $R^4$ 

### 酸性条件:

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- ANN Cheritan A HERELLE
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## Pfau-Plattner 英合成

从茚和重氯乙酸酯合成英。参见Buchner 扩环反应。

$$\begin{array}{c|c}
\hline
N_2CHCO_2R \\
\hline
-CO_2R \\
\hline
\end{array}$$

 $N_2$   $CO_2CH_3 \equiv CHCO_2CH_3 \equiv CHCO_2CH_3$ 

反应过程中重复化合物如同卡宾等价物

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# Pfitzinger 喹啉合成

碱性条件下邻氨基苯基乙酮酸(靛红酸)和α-亚甲基羰基化合物缩合生成 喹啉-4-羧酸。

$$CO_2$$
H
 $R^1$ 
 $R^2$ 
 $R^2$ 

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### Pictet-Gams 异喹啉合成

酰基化的胺甲基苯甲醇在 P2O5作用下生成异喹啉。

$$\begin{array}{c|c}
OH \\
HN \\
R
\end{array}$$

$$\begin{array}{c|c}
P_2O_5 \\
R
\end{array}$$

 $P_2O_5$  实际上以  $P_4O_{10}$  存在,如同金刚烷的结构。

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### Pictet-Spengler 异喹啉合成

### β-芳乙基胺和羰基化合物缩合再环化生成四氢异喹啉。

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### Pinner 合成

**腈转变为亚胺基醚,再继续转变为一个酯或脒。** 

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### Polonovski 反应

用乙酸酐等活化剂使叔氨氧化物重排产生 N,N-二取代的乙酰胺和一个醛。

$$R^1$$
  $CH_3CO)_2O$   $R^1$   $CH_2CI_2$   $R^1$   $CH_3CO)_2O$   $R^2$   $R^1$   $CH_2CI_2$   $R^2$   $R^2$ 

分子间过程也有可能:

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# Pomeranz-Fritsch 反应

从苯甲醛和胺基缩醛合成异喹啉。

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# North Life Trail City, At House Hall City, At

## Prilezhaev 反应

过酸对烯烃的环氧化(反应)。

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### Prins 反应

烯烃对甲醛的加成反应。

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# Pschorr

## Pschorr 闭环(反应)

分子内Gomberg-Bachmann 反应的扩展。

$$CO_2H$$
  $NANO_2$ ,  $HCI$   $CO_2H$   $NH_2$   $NH$ 

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## Pummerer 重排

用乙酸酐将亚砜转变为α-酰氧基碳醚。

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## Ramberg-Bäcklund 烯烃合成(反应)

 $\alpha$ -卤代砜用碱处理后生成烯烃。

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# ANNE CHERTICAL AND SELECTION OF THE PARTY OF

### Reformatsky 反应

有机锌试剂(从 $\alpha$ -卤代酯而来)对羰基的亲核加成(反应)。

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### Regitz 重氮盐合成

用对甲苯磺酰叠氮或甲磺酰叠氮合成2-重氮-1,3-二羰基或2-重氮-3-酮酯 类化合物。

只有一个羰基存在时,甲酸乙酯可作为活化剂[6-9]:

另外有可能的是,三唑中间体是能过烯醇和甲磺酰叠氯化物发生1,3-偶极环加 成反应而来的。

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### Reimer-Tiemann 反应

在碱性介质中从酚和氯仿合成邻甲酰基苯酚。

### 1. 生成卡宾:

$$Cl_3$$
C-H  $fast$   $H_2O$   $Cl_2$   $-Cl_2$   $Ccl_2$   $+Ccl_2$   $+Ccl_2$   $+Ccl_2$   $+Ccl_2$ 

### 2. 二氯卡宾加成和水解:

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## Reissert (醛合成)反应

从酰氯、喹啉和KCN制得相应的醛。

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## Riley 氧化 (SeO2 氧化) 反应

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# www.chenich 科機構供開作 利用 Gru

## 利用 Grubbs 和 Schrock 催化剂发生的环金属化(反应)

所有这三个催化剂都以"L<sub>n</sub>M=CHR"出现在下面的机理中。 由前催化剂产生的催化剂:

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normal detailed, Arthogodally the state of t

### Ritter 反应

腈和醇在强酸中生成酰胺。

通式:

$$R^1$$
-OH +  $R^2$ -CN  $\xrightarrow{H^+}$   $R^1$   $R^2$ 

例:

$$\rightarrow$$
 OH + H<sub>3</sub>C-CN  $\xrightarrow{\text{H}_2SO_4}$   $\xrightarrow{\text{N}}$ 

类似地:

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### Robinson 闭环(反应)

环己酮对甲苯乙烯基酮发生 Michael 加成反应后接着进行分子内 Aidol 缩合生成 六元环的  $\alpha$  ,  $\beta$ -不饱和酮。

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# Robinson-Schöpf 反应

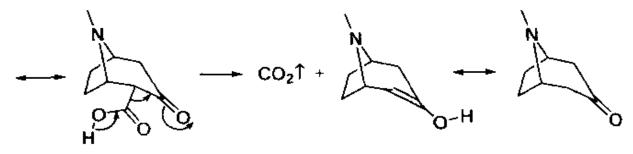
托品酮合成。

CHO + 
$$-NH_2$$
 +  $CO_2H$  +  $2CO_2\uparrow$  +  $2H_2O$ 

胺基 - 缩醛

$$HO_2C$$
 $O$ 
 $HO_2C$ 
 $O$ 
 $HO_2C$ 
 $O$ 
 $HO_2C$ 
 $O$ 

www.chemica philiphia.



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 $e_{i}^{-1}(t) \bullet ^{-1}$ 

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## Rosenmund 还原

用BaSO4/毒化钯催化剂将酰氯氢化还原为醛。催化剂未被毒化时,醛会继续

还为醇。

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ANATA LIBERTICA ANE MALE AND ANATA LIBERTICA AND ANATA LIBERTICA AND ANATA ANATA AND ANATA AND ANATA AND ANATA AND ANATA AND ANATA AND ANATA 
### Roush 烯丙基硼烷试剂

#### 酒石酸烯丙基硼酸酯不对称烯丙基化试剂

R-CHO + 
$$O_2$$
  $\dot{P}$   $\dot{P}$ 

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烯醇硅烷的 α- 羟基化 (反应)。

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### Rupe 重排

 $\alpha$ -末端炔基取代的叔醇在酸催化下重排为 $\alpha$ , $\beta$ -不饱和酮(非相应的 $\alpha$ , $\beta$ -不饱和 醛)。参见 Meyer-Schuster 重排。

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### Rychnovsky 多元醇合成

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## Sakurai 烯丙基化反应 (Hosomi-Sakurai 反应)

#### Lewis酸催化的烯丙基硅烷对亲核性碳的加成反应。

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### Mark chemistry Art To The Sandmeyer 反应

重氮盐和CuX反应得到芳基卤代物。

$$ArN_2^+ Y^- \xrightarrow{CuX} Ar-X X = Cl, Br, CN$$

例:

$$ArN_2^+ Cl^- \xrightarrow{CuCl} N_2\uparrow + Ar + CuCl_2 \longrightarrow Ar-Cl + CuCl_2$$

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### Sarett 氧化(反应)

用CrO。· Py 络合物将醇氧化为相应的羰基化合物。

分子内机理也有:

Collins 氧化、Jones 氧化和 Corey 的 PCC 及 PDC 氧化都有相同的机理。

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### Schiemann 反应 (Balz-Schiemann 反应)

从芳香胺合成芳香氟化物.

$$Ar - NH_{2} + HNO_{2} + HBF_{4} \longrightarrow ArN_{2}^{+} BF_{4}^{-} \longrightarrow Ar - F + N_{2} \uparrow + BF_{3}$$

$$HO^{-N}O \longrightarrow HBF_{4} \longrightarrow H_{2}O^{+} NO^{-} \longrightarrow H_{2}O + N=O^{+}$$

$$Ar \longrightarrow NH_{2} \longrightarrow H_{2}O + Ar \longrightarrow NO^{-} NO^{-} \longrightarrow ArN_{2}^{+} BF_{4} \longrightarrow N_{2}\uparrow + Ar^{+} + F-BF_{3} \longrightarrow Ar-F + BF_{3}$$

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### graph Company of Wittig 反应的修正

不稳定的叶立德和醛发生的 Wittig 反应生成(Z)-烯烃,而改进的 Schlosser 反应可得到(E)-烯烃。

苏式内盐的LiBr络合物

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### Schmidt 反应

用HN。将酮转变为酰胺。

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### Schmidt 三氯乙酰亚胺酯的苷化反应

Lewis酸促进的三氯乙酰亚胺酯和醇或酚发生苷化反应。

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### Scholl 反应

在 Friedel-Crafts 催化剂存在下,两芳基 要产生键联的氢消除生成一根芳基-芳 基键。参见 Friedel-Crafts 反应。

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环状亚胺和 $\beta$ -酮酯生成酮-哌啶化物。

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### Schotten-Baumann 反应

酰卤和醇或胺在碱性条件下发生酯化或酰胺化。

$$R = R^{1} - OH$$

$$R = R^{1} - OH$$

$$R^{1} $

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### Shapiro 反应

Shapiro反应是Bamford-Stevens反应的变异。前者用 Rli 和 RMgX等碱, 后者用 Na、NaOMe、BiH、NaH、NaNH2等碱。结果是,前者的反应给出一个动力学控制的少 取代烯烃产物,后者给出一个热力学控制的多取代烯烃产物。

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### Sharpless 不对称羟胺化(反应)

锇催化物的复和氧对烯烃的顺式加成,氯原子来源于:

手性配体 
$$R^1$$
  $K_2OsO_2(OH)_4$   $R$   $R^1$   $R^1$   $R^1$   $R^1$   $R^1$   $R^2$   $R^2$   $R^3$   $R^4$   $R$ 

催化循环:

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### Sharpless 不对称环氧化(反应)

用 Bu<sup>\*</sup> OOH、Ti(OPr<sup>\*</sup>)。和光学活性的酒石酸二乙酯对烯丙醇进行对映选择性环氧化反应。

活性催化物种为 [2];

过渡态为:

催化循环:

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## NAMES OF THE PROPERTY OF THE P

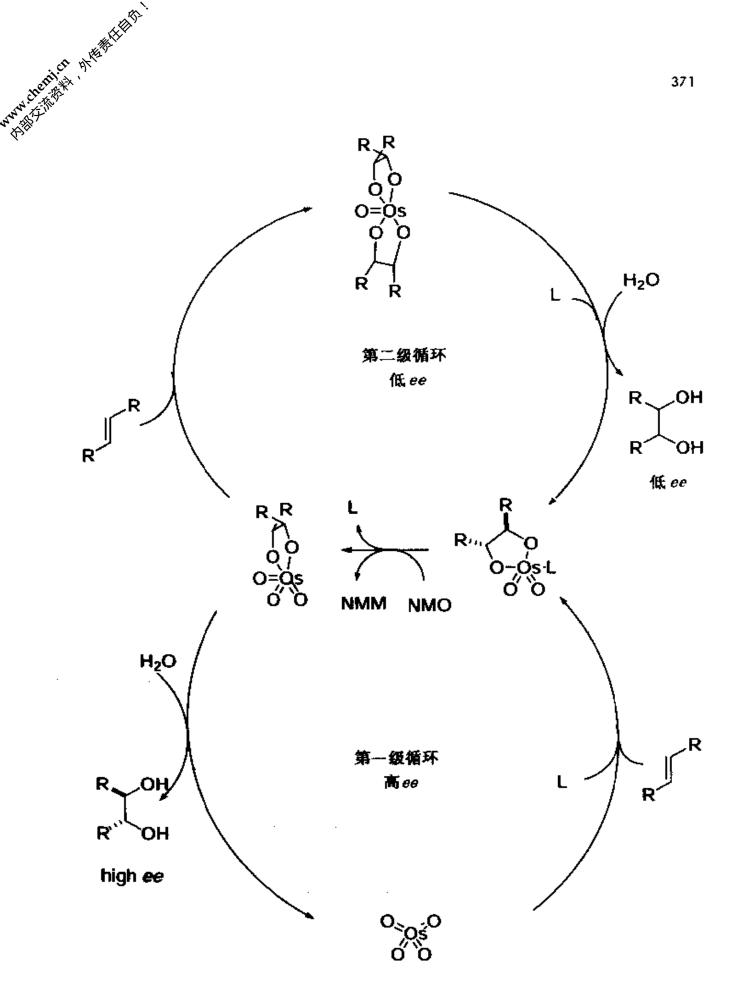
### Sharpless 二羟(基)化(反应)

在以金鸡纳生物碱为配体的铁催化下烯烃发生对映选择性顺式二羟(基)化反应。

包括镜杂环丁烷过程的多步机理看来比相应的协同[3+2]机理与实验数据更相符。

催化循环见下一页(P.371,第二级循环因低浓度烯烃存在会终止)。

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### Shi 不对称环氧化(反应)

用果糖产生的手性酮进行的不对称环氧化(反应):

催化循环:

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### Simmons-Smith 反应

用CH2I2和 Zn(Cu)对烯烃进行的环丙烷化。

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羧酸银和碘作用生成酯。但另一方面,当羧酸银和溴作用时产物是烷基溴(Hunsdiecker 反应, P. 202)。

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### Simonis 色酮环化(反应)

酚和 $\beta$ -酮酯用 $P_2O_5$  生成色酮。

 $P_2O_3$  实际上以一种类金刚烷结构的  $P_4H_{10}$ 形式存在。

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# Skraup 喹啉合成

苯胺、甘油、硫酸和氧化剂 (如PhNO<sub>2</sub>)生成喹啉。

还有一个可能的机理见Doebner-von Miller 反应 (P. 117)。

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# Smiles 重排

#### 通式:

例:

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# Sommelet 反应

用六亚甲基四胺将苄氯转化为相应的苯甲醛。

负氢的转移和六亚甲基四胺的开环可能是同步进行的:

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# Sommelet-Hauser (铵叶立德) 重排

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### Sonogashira 反应

Pd/Cu催化的有机卤和端基炔烃之间的交叉偶联反应。参见Castro-Stephens 反应。

$$R - X + = R' - R' - R' - R' - R'$$

$$Cul, Et_3N, rt$$

注意,Et<sub>3</sub>N也可还原Pd(II)到Pd(0)而同时Et<sub>3</sub>N被氧化为亚胺离子 [8];

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# Staudinger 反应

叠氮化物由 Ph<sub>3</sub>P/H<sub>2</sub>O 还原为胺:

$$X-N_3$$
  $\xrightarrow{PR_3}$   $X-N=N-N=PR_3$   $\xrightarrow{-N_2}$   $X-N=PR_3$   $\xrightarrow{-N_2}$   $X-N=PR_3$   $\xrightarrow{-N_2}$   $X-N=N-N=PR_3$   $\xrightarrow{-N_2}$   $X-N=N-N=PR_3$   $\xrightarrow{-N_2}$   $\xrightarrow{-N_2}$ 

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# Stetter 反应 (Michael-Stetter 反应)

从醛和 $\alpha,\beta$ -不饱和酮得到1,4-二羰基衍生物。噻唑啉錯催化剂相当于是一个安全的  $\mathrm{CN}^-$ 代用品。参见 Benzoin 缩合。

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# Stevens 重排

一个吸电子基团接在和氦相连的碳原子上的季铵盐用强碱处理给出重排的叔胺。

目前认可的自由基机理:

#### 原来认为的离子机理:

$$z \xrightarrow[R^2]{R^1} \longrightarrow z \xrightarrow[R^2]{R^1} \longrightarrow z \xrightarrow[R^2]{R^1}$$

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# Stieglitz 重排

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# Still-Gennari 膦酸酯反应

Horner-Emmons 反应用 KN(SiMe3)2 生成(E)-烯烃。

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# Kury Cir. (東京) Stille 偶联(反应)

Pd 催化的有机锡和有机卤、三氟磺酸酯等之间的交叉偶联反应。参见 Kumada 偶联反应 (P. 208)。

$$R-X + R^1-Sn(R^2)_3$$
  $Pd(0)$   $R-R^1 + X-Sn(R^2)_3$   $R-X + L_2Pd(0)$   $R-X + L_2Pd(0)$ 

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# Stille-Kelly 反应

用双锡试剂进行钯催化下二芳基卤代物的分子间交叉偶联反应。

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丁二酸二乙酯和羰基化合物在碱存在下的缩合反应。

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### Stork 烯胺反应

闭环反应的一种变异,反应中要用如吡咯烷之类较大体积的胺,这样能够使对甲基乙烯酮的共轭加成在两个可能的烯胺中从位阻较小的一面进攻。

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# Strecker 氨基酸合成

NaCN促进的醛和按缩合生成 $\alpha$ -氨基腈,后者水解为 $\alpha$ -氨基酸。

$$R^1$$
  $H$   $H_2N$   $R^2$   $ACOH$   $R^1$   $CN$   $R^1$   $CO_2H$   $R^1$   $CO_2H$   $R^1$   $CO_2H$   $R^1$   $R^2$   $R^2$   $R^1$   $R^2$   $R^2$   $R^1$   $R^2$   $R^2$ 

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# Ruzuki 偶联(反应)

钯催化下的有机硼烷和有机卤、三氟磺酸酯等在碱存在下发生的交叉偶联反应(若 无碱的活化作用金属转移化将受到阻碍)。催化循环见Kumada偶联反应(P. 234)。

$$R-X + R^{1}-B(R_{2})_{2} \xrightarrow{L_{2}Pd(0)} R-R^{1}$$

$$R-X + L_{2}Pd(0) \xrightarrow{\text{氧化加成}} R \xrightarrow{L} Pd \xrightarrow{X}$$

$$R^{1}-B(R_{2})_{2} \xrightarrow{\text{NaOR}^{3}} QR_{3}$$

$$R^{1}-B(R_{2})_{2} \xrightarrow{\text{NaOR}^{3}} R^{1}-B(R^{2})_{2}$$

$$R \xrightarrow{L} Pd \xrightarrow{X} R^{1}-B(R^{2})_{2} \xrightarrow{\text{PdA}} R$$

$$R^{3}O-B(R^{2})_{2} + Pd \xrightarrow{R^{1}} R$$

$$R^{3}O-B(R^{2})_{2} + Pd \xrightarrow{R^{1}} R$$

$$R^{2}-B(R^{2})_{2} + Pd \xrightarrow{R^{1}} R$$

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# Swern 氧化(反应)

用(COCl)2、DMSO为试剂和Et3N猝灭的方法将醇氧化为相应的羰基化合物。

$$\begin{array}{c} (COCI)_2, DMSO \\ (R^1 + R^2 + CI) CI (CI) CI ($$

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烷基氟硅烷被氧化为相应的醇,参见 Fleming 氧化(反应).

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# Tebbe 烯烃化(反应)[Petasis 烯基化(反应)]

Petasis 试剂 ( $Me_2$ Ti $Cp_2$ ,二甲基二茂钛) 与酮(醛) 也进行相似的烯烃化反应 [5] ,但两者的机理是完全不同的。

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# range — Ziegler 反应

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# Tiemann 重排

#### 从酰胺和羟胺而生成的偕胺肟和苯磺酰氯及水反应生成脲。

$$R^1$$
  $R^2$   $R^2$ 

$$R^{1}$$
 $N$ 
 $R^{2}$ 
 $R^{1}$ 
 $N$ 
 $R^{2}$ 
 $R^{1}$ 
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#### 与离去基(TOSO<sub>2</sub>Ph) 呈反式的取代基发生迁移

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# Tiff-

# Tiffeneau-Demjanov 重排

#### β-胺基醇经重氮化后发生碳正离子重排给出羰基化合物。

第一步,产生 N<sub>2</sub>O<sub>3</sub>

#### 第二步,胺转化为重氮盐

#### 第三步, 经重排扩环

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# rww.c.r.f.f.f.f. Tishchenko 反应

从醛和 Al(OEt); 得到相应的酯。Al(OEt); 是一个均相催化剂。

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δ-溴代烯烃发生自由基环化反应生成取代的四氢呋喃。

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## Ugi 反应

羧酸、异腈、胺和含氧化合物等四组份缩合反应生成肽。参见 Passerini 反应。

$$R-CO_2H+R^1-NH_2+R^2-CHO+R^3-N=C$$
 →  $R$   $R^1-N+C$  →  $R$   $R^1-N+C$   $R^3-N=C$   $R^3-N=C$   $R^1-N+C$   $R^3-N+C$   $R^3-N+$ 

$$\begin{array}{c|c} R-CO_2H & O & R^2 & H \\ \hline R & O & R^2 & H \\ \hline R^3 & & R^3 \\ \hline \end{array}$$

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## Ullmann 反应

芳基碘代物在Cu存在下的自偶联反应。

从Phi到PhCuI总的是一个氧化加成过程。

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## Vilsmeier-Haack 反应

Vilsmeier-Haack 试剂(氯代亚胺盐)是一个弱的亲电试剂,故 Vilsmeier-Haack 反应更适于和富电子的芳环和杂环反应。

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## 

#### 溴化氰和叔铵反应产生一个氨基氰和烷基卤代物。

$$\begin{array}{c} \begin{matrix} R \\ I \end{matrix} \\ R^{1} \end{matrix} \stackrel{R}{N} \begin{matrix} Br - CN \end{matrix} \qquad \begin{array}{c} \begin{matrix} CN \\ R^{1} \end{matrix} \stackrel{N}{N} \begin{matrix} R_{2} \end{matrix} \qquad + \quad Br - R \end{matrix}$$

溴化氰是一个两可进攻试剂。

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## von Richter 反应

硝基苯和氰化物反应在硝基的邻位产生羧基化。

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## Wacker 氧化(反应)

Pd催化下烯烃氧化为酮。

Pd(II) 再生:

Cu(II) 再生:

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# North Life Trail City, At House Hall City, At

## Wallach 重排

#### 氧化偶氮化合物经酸处理后生成对羟基偶氮化合物。

Ar 
$$N=N$$

Ar  $N=N$ 

Ar  $N=N$ 

Ar  $N=N$ 

Ar  $N=N$ 

Ar  $N=N$ 

Ar  $N=N$ 

OH

OH

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## Weinreb 酰胺

N-甲氧基-N-甲基酰胺。由于整合效应的关系,一个有机金属试剂的亲核加成只发生一次而给出酮,而正常的酰胺则发生二次加成而给出叔醇。

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## Weiss 反应

Cis-双环 [3.3.0] 辛 -3,7-二酮的合成。

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## Kuru Wenker 氮杂环丙烷合成

#### $\beta$ -胺基醇和硫酸给 $\beta$ -胺乙基硫酸,接着用碱处理后生成氮杂环丙烷。

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## Wharton 氧迁移反应

利用肼将α,β-环氧酮还原成烯丙醇。

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## www.chemich. Arthatelland

## Willgerodt-Kindler 反应

酮被转化为相应的硫代酰胺和/或铵盐。

$$\longrightarrow$$
 HN R +  $\longrightarrow$  SH

www.cherica splitter

稍有不同的机理也有可能:

在 Carmack 机理<sup>[5]</sup>中,最不寻常的羰基从亚甲基碳到羰基碳的迁移被认为是一个高度活泼的带有硫杂原子环的中间体所参与的复杂途径。氨硫化物是异构化的催化剂:

www.chenich 科格斯提供的

Mark Henich Arts Art Henry

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# KANGA Williamson 醚合成法

烷氧基由烷基卤烷基化后生成醚。

$$R^{1}-O = R^{1}O =$$

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水水水 cherica 特殊提供提介

## Wittig 反应

#### 羰基用膦叶立德变为烯烃。

$$Ph_{3}P + R^{1} \times R^$$

折叠状过渡态,不可逆的协同的过程

$$Ph_3PTO$$
  $Ph_3P=O + R^3$   $R^3$   $R^3$ 

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# RANGE CHERTICAL AND SELECTION OF THE PROPERTY OF THE PARTY OF THE PART

## [1,2]-Wittig 重排

#### 醚用RLi处理得酶。

自由基机理也有可能,因为自由基中间体已被检测到。

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## [2,3]-Wittig 重排

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# 

## Wohl-Ziegler 反应

以 AIBN 或光解下的 NBS 为催化剂用 NBS 进行自由基引发的烯丙位溴化反应。

烯丙基溴化(反应):

引发:

增长:

丁二酰亚胺自由基现在可以进入下一个自由基链反应的循环中。

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通过α-重氯酮和烯酮中间体发生的一碳同系化反应。

烯酮用水处理给出相应同系化的羧酸。

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## North Electricity Artifaction of the State o

#### Wurtz 反应

烷基卤经钠或镁金属处理后生成碳一碳键。

$$R-X$$
  $\xrightarrow{Na(0)}$   $R-R$  + NaX  
 $R-X$   $\xrightarrow{Na(0)}$   $R^ Na^+$  + NaX

离子机理

$$R^{-}$$
 $R = R + X^{-}$ 

自由基机理

$$R-X \xrightarrow{R^- Na^+} NaX + 2R \xrightarrow{\longrightarrow} R-R$$

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## Yamaguchi 酯化(反应)

2,4,6-三氯苯甲酰氯(Yamaguchi试剂)参与进行的酯化(反应)。

**DMAP** 

$$R_1$$
  $O$   $N_1$   $R_2$   $O$   $R_1$ 

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# Zaitsev 消除(反应)

E2 消除反应给出带更多取代基的烯烃。

Hofmann 消除在另一方面可生成带更少取代基的烯烃。

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# Zincke 反应

苯胺和1-(2,4-二硝基苯基) 吡啶氯化盐经开环反应生成 N-芳基或 N-烷基吡 啶盐。

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# Zinin 联苯胺重排(反应)(半联苯胺重排)

酸促进下二苯肼重排为4,4′-二胺基联苯(联苯胺)和2,4′-二胺基联苯。

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